

## **KA0801 Readme File**

### **Principal Investigator/Organization:**

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### **Ship Name:** Ka'imimoana

Call Sign: WTEU  
Country: United States  
Ship Owner: National Oceanic and Atmospheric Administration (NOAA)

### **Temporal Coverage:**

Cruise Start: February 2, 2008; Honolulu, Hawaii  
Cruise End: March 6, 2008; Kwajalein

**Shoreside support/Data Reduction:** Cathy Cosca, cathy.cosca@noaa.gov

**System Operators:** Tonya Watson (Survey Tech)

**Dataset ID/Location:** KA0801.csv ([www.pmel.noaa.gov/co2/uwpc02/](http://www.pmel.noaa.gov/co2/uwpc02/))

**Experiment Name:** Underway measurement of atmospheric and surface water pCO<sub>2</sub>

### **Geographical Bounds (+ E, - W for Longitude; + N, - S for Latitude):**

Westernmost Longitude: 167.613  
Easternmost Longitude: -154.911  
Northernmost Latitude: 8.788  
Southernmost Latitude: -13.267

### **Method Description:**

Equilibrator type/specifications: Showerhead, volume of ~0.5 L with a headspace of ~ 0.8 L.  
Water Flow rate: 3.5 L/minute  
Headspace gas flow rate: 60 ml/minute  
Measurement method: Infrared absorption of dried gas.  
CO<sub>2</sub> Sensor: Licor 6262, Serial # IRG3-1295  
Resolution/Uncertainty: 0.3 uatm for equilibrator measurements, 0.2 uatm for atmospheric measurements.

The general principle of instrumental design and operation are described in:

Feely, R.A., R. Wanninkhof, H.B. Milburn, C.E. Cosca, M. Stapp, and P.P. Murphy, A new automated underway system for making high precision pCO<sub>2</sub> measurements onboard research ships, *Analytica Chim. Acta*, 377, 185-191, 1998.

and

Wanninkhof and Thoning, Measurement of fugacity of Carbon Dioxide in surface water and air using continuous sampling methods, *Marine Chemistry*, 44, 189-205, 1993.

**Standard gases:**

Standard gases are supplied by NOAA's Climate Monitoring Diagnostics Laboratory in Boulder, CO, and are directly traceable to the WMO scale. Any value outside the range of the standards should be considered approximate, although the general trends should be indicative of the seawater chemistry.

Serial numbers and CO<sub>2</sub> concentrations for the cylinders used on this cruise:

CA03065	364.08
CA06867	421.13
CC115011	466.38
CA07519	543.65

**Sampling Cycle:**

The system runs a full cycle in approximately 112 minutes. The cycle starts with 4 standard gases, then measures 10 atmospheric samples followed by 60 surface water samples. Each new gas is flushed through the Licor Analyzer for 4 minutes prior to a 10 second reading from the analyzer during which the sample cell is open to the atmosphere. Subsequent samples of the same gas are flushed through the Licor Analyzer for 30 seconds prior to a stop-flow measurement.

**Units:**

All xCO<sub>2</sub> values are reported in parts per million by volume (ppmv) and fCO<sub>2</sub> values are reported in microatmospheres (uatm) assuming 100 % humidity at the equilibrators temperature.

**Calculations:**

The mixing ratios of ambient air and equilibrated headspace air are calculated by fitting a second-order polynomial through the hourly averaged response of the detector versus mixing ratios of the standards. Mixing ratios of dried equilibrated headspace and air are converted to fugacity of CO<sub>2</sub> in surface seawater and water saturated air in order to determine the fCO<sub>2</sub>. For ambient air and equilibrator headspace the fCO<sub>2a</sub>, or fCO<sub>2eq</sub> is calculated assuming 100% water vapor content:

$$fCO_{2a/eq} = xCO_{2a/eq}(P-pH_2O)\exp(B/11+2d/12)P/RT$$

where  $f_{CO_2a/eq}$  is the fugacity in ambient air or equilibrator,  $p_{H_2O}$  is the water vapor pressure at the sea surface temperature,  $P$  is the atmospheric pressure (in atm),  $T$  is the SST or equilibrator temperature (in K) and  $R$  is the ideal gas constant ( $82.057 \text{ cm}^3 \cdot \text{atm} \cdot \text{deg}^{-1} \cdot \text{mol}^{-1}$ ). The exponential term is the fugacity correction where  $B_{11}$  is the second virial coefficient of pure  $CO_2$

$$B_{11} = -1636.75 + 12.0408T - 0.032795T^2 + 3.16528E-5 T^3$$

$$\text{and } d_{12} = 57.7 - 0.118 T$$

is the correction for an air- $CO_2$  mixture in units of  $\text{cm}^3 \cdot \text{mol}^{-1}$  (Weiss, 1974).

The calculation for the fugacity at SST involves a temperature correction term for the increase of  $f_{CO_2}$  due to heating of the water from passing through the pump and through 5 cm ID PVC tubing within the ship. The water in the equilibrator is typically  $0.2 \text{ }^\circ\text{C}$  warmer than sea surface temperature. The empirical temperature correction from equilibrator temperature to SST is outlined in Weiss et al. (1982).

$$d \ln(f_{CO_2}) = (T_{eq} - SST) (0.0317 - 2.7851E-4 T_{eq} - 1.839E-3 \ln(f_{CO_2eq}))$$

where  $d \ln(f_{CO_2})$  is the difference between the natural logarithm of the fugacity at  $T_{eq}$  and SST, and  $T_{eq}$  is the equilibrator temperature in degrees C.

## File Format

	<b>COLUMN HEADER</b>	<b>DESCRIPTION</b>
1.	GROUP/SHIP:	PMEL/Ka'imimoana
2.	CRUISE_ID:	KA<Year><nth cruise of year>
3.	JD_GMT:	Decimal year day
4.	Date_DDMMYYYY	Date in the format DDMMYYYY
5.	TIME_HH:MM:SS:	GMT HH:MM:SS
6.	LAT_DEC_DEGREE: are in	Latitude in decimal degrees (negative values southern hemisphere).
7.	LONG_DEC_DEGREE: are	Longitude in decimal degrees (negative values in western latitudes).

8. xCO2W\_PPM: Mole fraction of CO2 (dry) in the headspace equilibrator at equilibrator temperature ( $T_{eq}$ ) in parts per million. Water comes from bow intake 2m below the water line.
9. xCO2A\_PPM: Mole fraction of CO2 in air in parts per million.
10. xCO2A\_INTERPOLATED\_PPM: xCO2atm\_ppm averaged linearly to match up with measurements xCO2eq\_ppm
11. PRES\_EQUIL\_hPa: Barometric pressure in the equilibrator
12. PRES\_SEALEVEL\_hPa: Barometric pressure in the atmosphere
13. EqTEMP\_C: Temperature in the equilibrator water.
14. SST(TSG)\_C: Temperature from the ship's bow intake.
15. SAL(TSG)\_PERMIL: Thermosalinograph salinity
16. fCO2W@SST\_uATM: Fugacity of CO2 in sea water in microatmospheres calculated as outlined in the DOE Handbook.
17. CO2A\_uATM: Fugacity of CO2 in air in microatmospheres
18. dfCO2\_uatm: Sea water fCO2 - air fCO2 in microatmospheres.
19. QC\_FLAG: Quality control flag  
 2 = Good value  
 3 = Questionable value  
 4 = Bad value
20. QC\_SUBFLAG: Descriptive quality control flag used when a value receives a "3" QC flag  
 1 = Outside of Standard Range  
 2 = Questionable/interpolated SST  
 3 = Questionable EQU temperature  
 4 = Anomalous  $\Delta T$  ( $E_{qT} - S_{ST}$ ) ( $\pm 1^\circ C$ )  
 5 = Questionable Sea Surface Salinity

6 = Questionable pressure  
7 = Low EQU gas flow  
8 = Questionable air value  
9 = Interpolated standard value  
10 = Other, see metadata

## References

- DOE (1994). Handbook of methods for the analysis of the various parameters of the carbon dioxide system in sea water; version 2. A.G. Dickson and C. Goyet, eds., ORNL/CDIAC-74.
- Feely, R.A., R. Wanninkhof, H.B. Milburn, C.E. Cosca, M. Stapp, and P.P. Murphy, A new automated underway system for making high precision pCO<sub>2</sub> measurements onboard research ships, *Analytica Chim. Acta*, 377, 185-191, 1998.
- Wanninkhof, R. and K. Thoning (1993) Measurement of fugacity of CO<sub>2</sub> in surface water using continuous and discrete sampling methods. *Mar. Chem.* 44(2-4): 189-205.
- Weiss, R. F. (1970) The solubility of nitrogen, oxygen and argon in water and seawater. *Deep-Sea Research* 17: 721-735.
- Weiss, R. F. (1974) Carbon dioxide in water and seawater: the solubility of a non-ideal gas. *Mar. Chem.* 2: 203-215.
- Weiss, R. F., R. A. Jahnke and C. D. Keeling (1982) Seasonal effects of temperature and salinity on the partial pressure of CO<sub>2</sub> in seawater. *Nature* 300: 511-513.

For questions or comments contact:

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